# The Effects of TiO<sub>2</sub> in the Performance of Mortar

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**Abstract.** Titanium dioxide (TiO<sub>2</sub>) is a material that contains photocatalysic which acts as self-cleansing agents on a material surface. In the application of TiO<sub>2</sub> in the civil engineering construction, it can be mixed with the fresh mortar known as TiO<sub>2</sub> mortar. Hence, a study on the characteristic strength of the TiO<sub>2</sub> mortar needs to be carried out. Moreover, the optimum dosage levels of the TiO<sub>2</sub> in the mortar can be determined. The characteristics of the TiO<sub>2</sub> were identified based on the compressive strength and flexural strength at 3, 7, 21 and 28 days. A total of five sets of specimens with different dosage levels were prepared and compared with the control specimen.

#### Introduction

Titanium Dioxide (TiO<sub>2</sub>) has received considerable attention in recent years as coating for concrete pavement [1,2]. The photocatalytic process of TiO<sub>2</sub> can be used as trap agent to absorb organic and inorganic air. It gives a promising benefit such as reduce the environmental impacts such as acidification, eutrophication, criteria air pollutants and smog formation. It is also practical for self-cleaning [3] and air-purifying concrete pavement [4]. For other material, the effect of TiO<sub>2</sub> on properties of polyethylene nano composites [5], self-cleaning glass [6] and as coating on residential window glass [7] have also been investigated. Effectiveness of TiO<sub>2</sub> in the perspective of environmental friendly building materials such as in the aspect of coating, cement, concrete, wallpaper, glass, porcelain and PVC profile has been investigated by Chen and Xu [8]. They found that, TiO<sub>2</sub> has excellent performance in improving human environment and facilitate the effectiveness of human health to achieve environmental protection. It also has great benefit in improving air quality inside a building [9]. A study on TiO<sub>2</sub> photocatalytic concrete for air purification with different percentage by weight with respect to the binder has been conducted by Husken et al. [10]. The effects of TiO<sub>2</sub> nanoparticles on flexural damage of self-compacting concrete (SCC) have been studied by Nazari and Riahi [11]. In their study, the TiO<sub>2</sub> nanoparticles has been replaced in concrete instead of cement with 4 % weight of SCC and it was found that the replacement of more than 4 % weight of SCC reduced the flexural strength of the SCC. In the production of building materials, the TiO<sub>2</sub> functions as an additive that significantly reduces the consumption of resources such as traditional energy. It is because TiO<sub>2</sub> particles crystallize in three forms, anatase, rutile, and brookite [12]. Anatase is a meta-stable that transforms into rutile at high temperature [13] and has photocatalyst semiconductor in environmental purification [14]. Since the TiO<sub>2</sub> contains an agent which acts as self-cleansing due to photocataclysic composition, it is useful in civil engineering construction. It can be used as a self-cleaning material in order to maintain the surface condition of the structure especially on reinforced concrete structure which is highly exposed to aggressive environment. The application of TiO<sub>2</sub> in such structure can be implemented by a generic study of its performance such as inclusion of TiO<sub>2</sub> in the mortar and concrete. However, the reviews on the inclusion of TiO<sub>2</sub> in both materials are still limited. In this paper, only the mixture of TiO<sub>2</sub> and fresh mortar at various dosage levels is presented. In the present study, only the compressive strength and the flexural strength were identified at 3, 7, 21 and 28 days of harden TiO<sub>2</sub> mortar specimens. The results were then compared with the control specimen and the strength effectiveness of the TiO<sub>2</sub> mortar specimens was identified.

### **Experimental Programme**

The physical properties of the  $TiO_2$  are solid state (powder), slightly odour and white colour. The chemical properties of the  $TiO_2$  are density of  $4.05~\rm g/cm^3$  and  $1800^0$ c melting point. In the preparation of the  $TiO_2$  mortar, the mortar was designed for different proportions by weight of water: cement: fine aggregate and it was 0.5: 1: 3, respectively. Then, 5 % of silica fume based on cement weight was added to the mortar mix to improve the workability of the mortar. Table 1 shows the dosage levels of the  $TiO_2$  used in the mortar mix. For control specimen, no  $TiO_2$  and silica fume were added to the mortar mix. Two sizes of  $TiO_2$  mortar were prepared,  $100~\rm mm~x~100~mm~x~10$ 

Mixture	Cement (%)	Titanium Dioxide (%)	Silica Fume (%)
Control	100	0	0
TDM 1	67	0.28	5
TDM 2	65	0.30	5
TDM 3	63	0.32	5
TDM 4	61	0.34	5
TDM 5	59	0.36	5

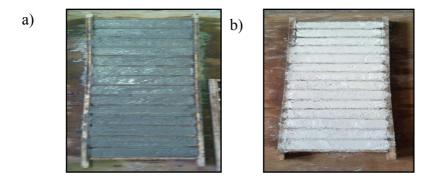


Fig. 1. Fresh mix in the moulds a) control prisms b) TiO<sub>2</sub> mortar prisms

#### Results and Discussion.

Compressive Strength Results. Fig. 2a shows the results of compressive strength of hardened control specimen and TiO<sub>2</sub> mortar at 3, 7, 21 and 28 days. Five dosage levels of TiO<sub>2</sub> were mixed in the mortar and designated as TDM 1 to 5. From the graph, the compressive strength of TiO<sub>2</sub> mortar for all dosage levels of TDM 1 to 5 increases as the age of the specimen increases. For TDM 1, which contains 0.28 % of TiO<sub>2</sub>, the compressive strength is lesser than TDM 2 which contains 0.30 % of TiO<sub>2</sub> at the age of 3 to 21 days. However, the compressive strength increases when it reaches the optimum age of 28 days. It is found that the compressive strength of TDM 1 is 23.32 N/mm<sup>2</sup> and TDM 2 is 21.97 N/mm<sup>2</sup>. The compressive strength of TDM 3 which contains 0.32 % of TiO<sub>2</sub> drastically dropped halves than TDM 2 at the age of 3 days. It indicates that the increase of TiO<sub>2</sub> in the mortar mix did not increase the strength at early age of the specimen. Similar case occurred for TDM 4 and TDM 5 which contains 0.34 % and 0.36 % of TiO<sub>2</sub>. At optimum age of 28 days, it is found that the compressive strength reduces when the TiO<sub>2</sub> increases. From the Fig. 2a, it is also

found that the compressive strength of TiO<sub>2</sub> mortar is lesser than control specimen. The inclusion of TiO<sub>2</sub> in mortar increased the porosity of the mortar and hence reduces the strength of the mortar. In overall, it indicates that the inclusion of TiO<sub>2</sub> did not increase the compressive strength of the mortar especially at the optimum age of 28 days. However, the compressive strength of TiO<sub>2</sub> mortar for specimen TDM 2 at the age of 21 days increases more than the control specimen. The compressive strength effectiveness of the TiO<sub>2</sub> mortar is presented in Fig. 2b. It indicates that the TDM 2 which contains the inclusion of 0.30 % TiO<sub>2</sub> produced more strength effectiveness than the other specimens which have different inclusion of TiO<sub>2</sub> dosage levels at early age of 3 to 21 days. However, the strength effectiveness reduced to 3.3 N/mm² when it reached 28 days. It indicates that the optimum inclusion of TiO<sub>2</sub> in mortar is 0.30 %. If higher percentage of TiO2 is used, it reduced the strength of its combination. The compressive strength effectiveness can be used to determine the difference in the strength between the control specimen and the mortar with the inclusion of TiO<sub>2</sub>. From the Fig. 2b, the TDM 5 has lower compressive strength effectiveness compared to other specimens. In overall, the compressive strength effectiveness of TiO<sub>2</sub> mortar increases when it reaches the age of 21 days. However, it reduces when reaches to optimum age of 28 days.

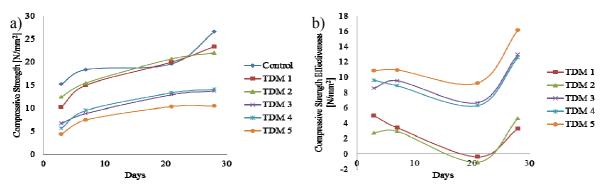


Fig. 2 a) Compressive strength of the control and TiO<sub>2</sub> mortar; b) Compressive strength effectiveness of the various dosage levels of TiO<sub>2</sub> mortar

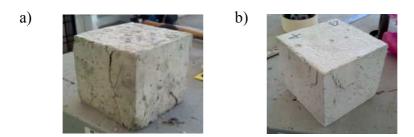


Fig. 3. Condition of the cube surface a) control cube b) mortar with the inclusion of TiO<sub>2</sub>

Although the TiO<sub>2</sub> seemingly did not increase the performance of the mortar in terms of its compressive strength, at the age 21 days, the inclusion of TiO<sub>2</sub> at low dosage level such as 0.28 % and 0.30 % TiO<sub>2</sub> in the mortar is found to increase when compared to the control specimen. This depicts that it is still practical to be used if the inclusion of other material in the composition of the mortar can be made. At the same time, the TiO<sub>2</sub> role as self-cleaning agent in the composition can still be maintained. It can be depicted in Fig. 3 that the surface of the TiO<sub>2</sub> mortar cube is brighter or cleaner than the control cube. Hence, the inclusion of TiO<sub>2</sub> in the mortar composition can be used as decorative materials.

Flexural Strength Results. The flexural strength of control specimen and various dosage levels of TiO<sub>2</sub> mortar specimen are depicted in Fig. 4a. It indicates that the flexural strength of the control specimen increases as the age of the specimen increases. The average flexural strength of the control mortar increased from 4.45 N/mm<sup>2</sup>, 6.03 N/mm<sup>2</sup>, 6.73 N/mm<sup>2</sup> and 7.86 N/mm<sup>2</sup> for age 3, 7, 21 and 28 days, respectively. The flexural strength of TDM 1 at the age 3 days was 5.16 N/mm<sup>2</sup> and

increased to 7.40 N/mm² when reached the optimum age of 28 days. It indicates that the harden  $TiO_2$  mortar prism specimen when subjected to three point loading increases the flexural strength as the age of the specimen increases. However, the inclusion of the  $TiO_2$  in the mortar mix is found to reduce the flexural strength as the dosage level of the  $TiO_2$  increases. It can be depicted that in Fig. 4a where at optimum age of 28 days, the flexural strength of TDM 1 (0.28 %  $TiO_2$ ), TDM 2 (0.30 %  $TiO_2$ ), TDM 3 (0.32 %  $TiO_2$ ), TDM 4 (0.34 %  $TiO_2$ ) and TDM 5 (0.36 %  $TiO_2$ ) were 7.40 N/mm², 6.94 N/mm², 5.90 N/mm², 5.16 N/mm² and 4.18 N/mm², respectively. It seemingly indicates that the higher the inclusion of  $TiO_2$  in the mortar mix did not improve the performance of the flexural strength. The flexural strength effectiveness of the  $TiO_2$  mortar at various dosage levels is depicted in Fig. 4b. It indicates that the inclusion of 0.28 %  $TiO_2$  and 0.30 %  $TiO_2$  designated as TDM 1 and TDM 2 have better strength effectiveness than the other prisms that has other inclusion of  $TiO_2$ . It can be seen especially at the optimum age of the prisms. In overall, the flexural strength effectiveness of  $TiO_2$  is lesser than the compressive strength effectiveness as discussed in preceding section. However, it is confirmed that 0.3 % of  $TiO_2$  is the optimum dosage level can be used to improve the strength of the mortar.

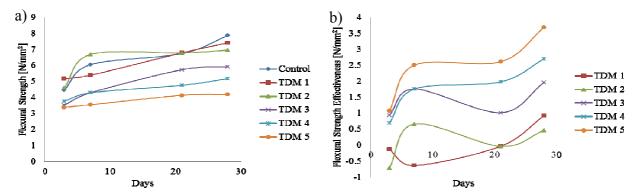


Fig. 4 a) Flexural strength of the control and TiO<sub>2</sub> mortar; b) Flexural strength effectiveness of the various dosage levels of TiO<sub>2</sub> mortar

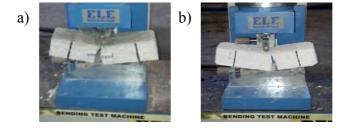


Fig. 5. Prism failure of hardened specimens a) control specimen b) TiO<sub>2</sub> mortar with 0.34% TiO<sub>2</sub>

Fig. 5 shows the failure of the prism specimens when subjected to monotonic load where the specimens fail almost halves for control specimen and  $TiO_2$  mortar specimen. From the observation of the appearance of the surface of the specimens, it is found that the specimens of  $TiO_2$  mortar are brighter than control prisms. It is due to the composition of photocataclysic in the  $TiO_2$  which affect to the surface of the specimen. Although the flexural strength of the  $TiO_2$  mortar is lesser than control specimen, this study is useful. It is because the review on the application of  $TiO_2$  in the civil engineering construction is still limited. Hence, this study can be used as a milestone or reference to other researchers who are interested in the application of  $TiO_2$  in the mortar.

### Conclusion

In overall, the comparison between TiO<sub>2</sub> mortar specimen and control specimen indicates that the strength of control specimen is higher than TiO<sub>2</sub> mortar. Hence, the inclusion of TiO<sub>2</sub> did not

increase the performance of mortar such as compressive strength and flexural strength. Although the strength of the  $TiO_2$  is lesser than control specimen, this study is vital as the inclusion of  $TiO_2$  in the mortar can be used for the structure appearance where it gains brighter feature of the structure's surface. It can be concluded that the optimum dosage level that can be used to improve the strength of the mortar is 0.3% of  $TiO_2$ . For future research, the  $TiO_2$  mortar will be applied in the reinforced concrete structure to investigate the performance of the surface that relates to the self-cleaning structure.

## References

- [1] M.M. Hassan, "Life cycle assessment of titanium dioxide coatings" Construction Research Congress, pp. 836-845, 2009.
- [2] M.M. Hassan, H. Dylla, L.N. Mohamad and T. Rupnow, "Evaluation of the durability of titanium dioxide photocatalyst coating for concrete pavement," Construction and Building Materials, vol. 24, pp. 1456-1461, 2010.
- [3] A. Fujishima, T.N. Rao and D.A. Tryk, "Titanium dioxide photocatalysis," Journal of Photochmistry and Photobiology C: Photochemistry Reviews vol. 1, pp. 1-21, 2000.
- [4] M. Schmitt, H. Dylla, M.M. Hassan, L.N. Mohammad, T. Rupnow and E. Wright "Impact of mixed nitrogen dioxide (NO.d2) and nitrogen oxide (NO) gases on titanium dioxide photodegradation of NO/dx," Transportation and Development Institute Congress, 2011.
- [5] V.G. Nguyen, H. Thai, D.H. Mai, H.T. Tran, D.L. Tran and M.T. Vu "Effect of titanium dioxide on the properties of polyethylene/ TiO<sub>2</sub> nanocomposites," Composites Part B, vol. 45, p.1192-1198, 2013.
- [6] E.I. Cedillo-Gonzalez, R. Ricco, M. Montorsi, M. Montorsi, P. Falcaro and C. Siligardi "Self-cleaning glass prepared from a commercial TiO<sub>2</sub> nano-dispersion and its photocatalytic performance under common anthropogenic and atmospheric factors," Building and Environment, vol. 71, pp. 7-14, 2014.
- [7] H. Babaizadeh and M. Hassan, "Life cycle assessment of nano-sized titanium dioxide coating on residential windows," Construction and Building Materials, vol. 40, p.314-321, 2013.
- [8] F. Chen and Y.Z. Xu, "The application of titanium dioxide of environment-friendly building materials," Applied Mechanics and Materials, vol. 174-177, p.767-770, 2012.
- [9] A. Awadalla, M.F. Mohd Zain, A.A.H. Kadhum and Z. Abdalla, "Titanium dioxide as photocatalyses to create self-cleaning concrete and improve indoor air quality," International Journal of the Physical Science, vol. 6 (29), p. 6767-6774, 2011.
- [10] G. Husken, M. Hunger and H.J.H. Brouwers, "Experimental study of photocatalytic concrete products for air purification," Building and Environment, vol. 44, pp. 2463-2474, 2009.
- [11] A. Nazari and S. Riahi, "The effects of TiO<sub>2</sub> nanoparticles on flexure damage of self-compacting concrete," International Journal of Damage Mechanics, vol. 20, pp. 1049-1072, 2010.
- [12] D.J. Osborn, "Quantification of NOx reduction via nitrate accumulation on a TiO<sub>2</sub> photocatalytic concrete pavement," Master Thesis, Louisiana State University, 2012.
- [13] L.R. Znaidi, J. Seraphimova, C. Bocquet, Colbeau-Justin and C. Pommier, "Continuous 35 process for the synthesis of nanosize TiO<sub>2</sub> powders and their use as photocatalysts," Material Research Bulletin, vol. 36, pp. 811-825, 2001.
- [14] S. Bilmes, P. Mandelbaum, F. Alvarez and N. Victoria, "Surface and electronic structure of titanium dioxide photocatalyst," Journal of Physical Chemistry B., vol. 104, pp. 9851-9858, 2000.