Residual stress study in oxide scale obtained on high temperature oxidation of AISI 430 stainless steel

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Abstract: The objective of this work was to investigate high temperature oxidation behavior of AISI 430 stainless steel, which was proposed to use as interconnector in the planar solid oxide fuel cells (SOFCs). The oxidation of the alloy has been conducted at 700°C, 800°C and 900°C for 12h-96h by thermal gravimetric analysis (TGA) system. The oxide surface morphology, cross-section microstructure and the chemical composition of the oxide scales were performed by FEG-SEM and EDX. The X-ray diffraction (XRD) was used to identify the oxide phases formed on the alloy and to determine the residual stress in the scale. It has been found that the oxide scale composed of a inner $\text{Cr}_2\text{O}_3$ layer and an outer $\text{Mn}_{1.5}\text{Cr}_{1.5}\text{O}_4$ layer. The residual stresses in both oxide layers are compressive and the residual stress evolutions in the two layers are different according the oxidation temperature.

1. Introduction

Solid oxide fuel cells (SOFCs) are electrochemical devices that convert fuel and oxidant gases into electricity. The reduction of the SOFCs operating temperature from 1000°C to about 800°C in recent years widens the materials choice for stack component such as the interconnects [1-5]. Ferritic stainless steels have many advantages, such as good electrical conductivity, appropriate thermal expansion coefficient and low cost, are among the most promising candidates as the SOFC interconnects.

Many oxidation studies have been done for ferritic steel at high temperature, but all these works were mainly focused on oxidation kinetics. However, in the oxide scale systematically accompanied the development of growth stresses during isothermal oxidation and thermal stress during cooling, which also limit the lifetime of the alloy. Although some works have been done for Ni-Cr alloy found that the oxide growth stresses are negligible [6-8], seldom works have been done for ferritic stainless steels. The aim of the present study is to investigate the residual stresses in the oxide scale after high temperature oxidation of AISI 430 ferritic stainless steel.

2. Experimental method

The AISI 430 ferritic stainless steel used in this work was supplied by PX Precimet SA, the chemical composition is listed in Table 1. For the thermal gravimetric analysis (TGA), the samples were cut to a dimension of 10mm x 10mm x 1mm, and a small hole of 0.8 mm in diameter near an edge to hang them by Platinum cricket in the TGA system. All samples polished by SiC paper followed by silicon paste (up to 2μm) to make sure the samples have the same surface condition. The TGA (model SETARAM 92-16.18) was employed to study the oxidation kinetics, the experiments have been carried out at 700°C, 800°C and 900°C for 12h-96h in artificial air. Each oxidation condition has been preformed with at least 2 samples, the results were reproducible.

| Chemical composition of AISI 430 stainless steel (weight %) |
|-----------------|-----|-----|-----|-----|-----|-----|-----|
| Fe   | Cr   | Mn   | Si   | P    | S    | C    |
| Bal. | 16-18| < 1.00| < 1.00| < 0.04| < 0.03| < 0.08 |
After oxidation, the surface morphology and composition of the oxide scale and the cross-section microstructure were examined by a FEG-SEM (ZEISS SUPRA 55VP) equipped with an EDX system (Energy-dispersive X-ray spectroscopy). Grazing Incidence X-ray diffraction (GIXRD) technique has been used for oxide phase identification with Panalytical X’Pert under Copper radiation (\(\lambda_{K\alpha}=0.154\text{nm}\)) with a fixed incident angle of 2°, and the analyzed depth is about 3\(\mu\text{m}\).

For residual stress analysis in oxide layers, European standard NF EN 15305 (version of April 2009) [9] has been applied with XRD method [10]. The \{104\} peaks for \(\text{Cr}_2\text{O}_3\), \{311\} peaks for \(\text{Mn}_{1.5}\text{Cr}_{1.5}\text{O}_4\) and \{110\} peaks for substrate were selected to determine the residual stress level because these peaks provide sufficient intensities, and more than 13 peaks with different Psi angles (varying from -60° to +60°) have been recorded. Because of high resolution configuration of XRD system, the peak position error determination is less than 0.005% whatever the 20 position.

3. Results and discussion

3.1. Oxidation kinetics

Fig. 1 shows the mass gain curve as a function of isothermal oxidation duration at 700°C, 800°C and 900°C, it is clear that all the oxidation rates obeyed a parabolic law [11]:

\[
(\Delta M/S)^2 = k_p t
\]

where \(\Delta M\) is the weight change, \(S\) is the surface area of the sample, \(k_p\) is the parabolic rate constant and \(t\) is the oxidation time. The parabolic rate constants calculated from the kinetic curves are listed in Table 2. According to Wagner theory, a parabolic oxidation rate indicates the formation of a protective scale acting as a diffusion barrier and the oxidation kinetics is controlled by ions diffusion through the oxide scale [12]. Fig. 2 shows a linear relationship between \(\log k_p\) and 1/T at different oxidation temperature, and the activation energy is 279.9 KJ/mol comparable to that obtained by M. Palcut et al. [13].

![Fig. 1. Weight gain curves of AISI 430 steel during exposure at 700°C, 800°C and 900°C](image1)

![Fig. 2. Log \(k_p\) of AISI 430 as a function of (1000/T) at 700°C, 800°C and 900°C](image2)

<table>
<thead>
<tr>
<th>Temperature</th>
<th>(k_p) (g(^2)cm(^{-4})s(^{-1}))</th>
</tr>
</thead>
<tbody>
<tr>
<td>700°C</td>
<td>(3.678 \times 10^{-15})</td>
</tr>
<tr>
<td>800°C</td>
<td>(7.527 \times 10^{-14})</td>
</tr>
<tr>
<td>900°C</td>
<td>(1.357 \times 10^{-12})</td>
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</tbody>
</table>
3.2. Microstructure and composition of the scales

The scale composition and microstructure were studied at room temperature by GIXRD and by FEG-SEM/EDX. Fig. 3 shows the GIXRD pattern of AISI 430 stainless steel oxidized for 96h at 700°C, 800°C and 900°C, which indicates the growth of Cr₂O₃ (JCPDS 38-1479) and spinel-type Mn₁.₅Cr₁.₅O₄ (JCPDS 33-0892) on the alloy surface. As the oxide scale is very fine, the substrate peaks can also be observed in the XRD pattern. And there are no obvious texture effects, because all XRD intensities correspond to theoretical proportions.

![GIXRD patterns for AISI 430 oxidized at 700°C, 800°C and 900°C for 96h in air](image)

Fig. 3. GIXRD patterns for AISI 430 oxidized at 700°C, 800°C and 900°C for 96h in air

Fig. 4 displays the surface morphology of scale on the substrate oxidized for 48h at 900°C. With the EDX analyses, we observed that the scale is mainly consisted of Cr, Mn and oxygen. Observation of the cross section by SEM showed that the oxide scale consisted of two layers, EDX analyses indicates that the outer layer is Mn₁.₅Cr₁.₅O₄ and the inner layer is Cr₂O₃ (Fig. 5). In the meantime, this can also explain that the relative high intensity of Mn₁.₅Cr₁.₅O₄ peaks occur due to presence of the Mn₁.₅Cr₁.₅O₄ phase in outer part of the scale but not to the higher proportion of Cr₂O₃ in the scale. At the beginning of the oxidation, the oxidation was mainly controlled by the outward diffusion of Cr³⁺, and a protective Cr₂O₃ oxide layer formed. After that the oxidation was mainly controlled by the outward diffusion of Mn³⁺ and a outer spinel layer formed, because Mn ion diffuses faster than Cr³⁺ via Cr³⁺ lattice in Cr₂O₃. Since the percentage composition of Mn in the alloy is small, then the oxidation was controlled by the diffusion of Cr³⁺ again, and the Cr₂O₃ layer thicken, the thickness of the oxide layers after oxidation can be found in the Fig. 7 and Fig. 8.

![SEM surface morphology of AISI 430 oxidized at 900°C for 48h](image)

Fig. 4. SEM surface morphology of AISI 430 oxidized at 900°C for 48h

3.3. Residual stress analysis

The in plane residual stresses (RS) in the chromia layer, the spinel layer and the substrate were determined after oxidation at room temperature. The Young’s modulus and Poisson’s ratio used for the residual stress calculation were 280MPa and 0.29 for Cr₂O₃ [8,10], 250MPa and 0.27
for Mn$_{1.5}$Cr$_{1.5}$O$_4$ [14], 220MPa and 0.28 for substrate, respectively. The $d$ vs. $f(\sin^2\Psi)$ plots are displayed in Fig. 6. For oxide phases, the anisotropic factor has been fixed to be 1 because of the lack of information in bibliography.

Fig. 5. Cross section after oxidation at 900°C for 48h and EDX analysis on oxide scale

The results of the RS in oxide scale after oxidation from 12h to 96h at 900°C were shown in Fig. 7, this indicates that the RS are compressive, and as the time goes on, the changing trends of the RS in the spinal layer and the chromia layer are the same at 900°C. From 12h to 24h and 48h to 96h the RS increased, especially in the spinal scale, which indicates that the growth stress plays an important role in RS for AISI 430, which is in contrary to the previous studies of RS in Cr$_2$O$_3$ for Ni-Cr alloy [6-8]. And from 24h to 48h the RS decreased, indicates stresses are relaxed as the oxide scale thicken. The stress relief may arise mainly due to the undulation of the scale and the substrate [Fig. 5].

Fig. 6 The $d = f(\sin^2\Psi)$ plot of oxide layers after oxidation for 96h at 900°C
Table 3 lists the RS in substrate after oxidation at 900°C, in Fig. 3 the intensity of the (110) diffraction peak at 900°C is low because it is the pattern of GIXRD. The stresses are small and compressive, there is no much difference between the RS in the substrate with increasing oxidation time. It’s in agreement with the result obtained by A.M. Huntz et al. [7] on Nickel. After undulation of the substrate, the stresses are still present may due to good adherence between the oxide scale and the substrate.

Table 3 Residual stress in substrate after oxidation at 900°C

<table>
<thead>
<tr>
<th>Condition</th>
<th>RS in substrate (MPa)</th>
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<tbody>
<tr>
<td>900°C 12h</td>
<td>− 80 ± 10</td>
</tr>
<tr>
<td>900°C 24h</td>
<td>− 90 ± 10</td>
</tr>
<tr>
<td>900°C 48h</td>
<td>− 95 ± 15</td>
</tr>
<tr>
<td>900°C 96h</td>
<td>− 75 ± 20</td>
</tr>
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Fig. 7 Residual stress in oxide scale after oxidation at 900°C for 12h-96h

Fig. 8 shows the effect of oxidation temperature on the RS in chromia layer and in spinal layer. It can be seen that the oxidation temperature has different effect on two layers. For the chromia layer, after 96h oxidation from 700°C to 900°C, the RS is decreased. But for the spinal layer, the compressive RS increased obviously. So the influences of the temperature on RS in the two layers are definitely different. However, seldom previous theoretical and experimental studies considered this difference, just saw the oxide layers as a whole during calculate and measure the RS. The reason for the RS difference between the two layer are still unknown, more studies need to be done to find the RS relationship between two oxide layers at different oxidation temperature.

4. Conclusions

The oxidation of AISI 430 stainless steel have been carried out at 700°C, 800°C and 900°C, the basic conclusions from present work can be made as follows:
(1) The oxide scale is consisted of a inner Cr2O3 layer and an outer Mn1.5Cr1.5O4 layer, and the activation energy is 279.9 KJ/mol;
(2) The residual stresses in both oxide layers are compressive. The growth stress plays an important role in residual stress in oxide scale for AISI 430;
(3) The influences of the temperature on residual stress in the two layers are different, which was unnoticed before.
References